

# **Pink Detergent**

#### Nowchem

Version No: 1.5 Safety Data Sheet according to WHS and ADG requirements Chemwatch Hazard Alert Code: 3

Issue Date: **05/10/2016**Revised Date: **05/10/2021**L.GHS.AUS.EN

### SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

## **Product Identifier**

Product name	Pink Detergent
Synonyms	Not Available
Other means of identification	Not Available

#### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses Pink washing up detergent for hand dishwashing applications.

### Details of the supplier of the safety data sheet

Registered company name	Nowchem
Address	112A Albatross Road NSW Australia
Telephone	(02) 4421 4099
Fax	(02) 4421 4932
Website	www.nowchem.com.au
Email	sales@nowchem.com.au

## Emergency telephone number

Association / Organisation	Nowchem
Emergency telephone numbers	(02) 4421 4099
Other emergency telephone numbers	0413 809 255

## **SECTION 2 HAZARDS IDENTIFICATION**

## Classification of the substance or mixture

HAZARDOUS CHEMICAL. NON-DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

## CHEMWATCH HAZARD RATINGS

	Min	Max	
Flammability	1		
Toxicity	0		0 = Minimum
Body Contact	3		1 = Low 2 = Moderate
Reactivity	1		3 = High
Chronic	0		4 = Extreme

Poisons Schedule	Not Applicable	
Classification [1]	Classification [1] Skin Corrosion/Irritation Category 2, Serious Eye Damage Category 1	
Legend: 1. Classified by Chemwatch; 2. Classification drawn from HSIS; 3. Classification drawn from EC Directive 1272/2008 - Annex VI		

#### Label elements

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# **Pink Detergent**

GHS label elements



SIGNAL WORD

DANGER

#### Hazard statement(s)

H315	Causes skin irritation.
H318	Causes serious eye damage.

## Precautionary statement(s) Prevention

P101	If medical advice is needed, have product container or label at hand.	
P102	Keep out of reach of children.	
P103	03 Read label before use.	
P280 Wear protective gloves/protective clothing/eye protection/face protection.		

### Precautionary statement(s) Response

P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P310	P310 Immediately call a POISON CENTER or doctor/physician.	
P362	Take off contaminated clothing and wash before reuse.	
P302+P352	IF ON SKIN: Wash with plenty of soap and water.	
P332+P313 If skin irritation occurs: Get medical advice/attention.		

#### Precautionary statement(s) Storage

Not Applicable

### Precautionary statement(s) Disposal

Not Applicable

## SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

## Substances

See section below for composition of Mixtures

## Mixtures

CAS No	%[weight]	Name
8046-53-5	<10	(linear)alkylbenzenesulfonic acid, sodium salts
8051-30-7	<10	diethanolamine cocoate
68891-38-3	<10	sodium linear-(C12-14)alkyl ether sulfate
2634-33-5	<1	1,2-benzisothiazoline-3-one

## **SECTION 4 FIRST AID MEASURES**

# Description of first aid measures

Description of most aid measures		
Eye Contact	If this product comes in contact with the eyes:  Immediately hold eyelids apart and flush the eye continuously with running water.  Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.  Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.  Transport to hospital or doctor without delay.  Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.	
Skin Contact	If skin contact occurs:  Immediately remove all contaminated clothing, including footwear.  Flush skin and hair with running water (and soap if available).  Seek medical attention in event of irritation.	
Inhalation	<ul> <li>If fumes, aerosols or combustion products are inhaled remove from contaminated area.</li> <li>Other measures are usually unnecessary.</li> </ul>	
Ingestion	<ul> <li>Immediately give a glass of water.</li> <li>First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.</li> </ul>	

## Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

## **SECTION 5 FIREFIGHTING MEASURES**

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## Extinguishing media

- ▶ Water spray or fog.
- Foam.
- ► Dry chemical powder.
- ► BCF (where regulations permit).
- ▶ Carbon dioxide.

### Special hazards arising from the substrate or mixture

Fire Incompatibility	<ul> <li>Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result</li> </ul>

### Advice for firefighters

Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear full body protective clothing with breathing apparatus.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> <li>Use water delivered as a fine spray to control fire and cool adjacent area.</li> <li>Avoid spraying water onto liquid pools.</li> <li>DO NOT approach containers suspected to be hot.</li> <li>Cool fire exposed containers with water spray from a protected location.</li> <li>If safe to do so, remove containers from path of fire.</li> </ul>
Fire/Explosion Hazard	<ul> <li>▶ Combustible.</li> <li>▶ Slight fire hazard when exposed to heat or flame.</li> <li>▶ Heating may cause expansion or decomposition leading to violent rupture of containers.</li> <li>▶ On combustion, may emit toxic fumes of carbon monoxide (CO).</li> <li>▶ May emit acrid smoke.</li> <li>▶ Mists containing combustible materials may be explosive.</li> <li>Combustion products include; carbon dioxide (CO2) other pyrolysis products typical of burning organic materialMay emit corrosive fumes.</li> </ul>
HAZCHEM	Not Applicable

### **SECTION 6 ACCIDENTAL RELEASE MEASURES**

## Personal precautions, protective equipment and emergency procedures

See section 8

### **Environmental precautions**

See section 12

## Methods and material for containment and cleaning up

Minor Spills	<ul> <li>Remove all ignition sources.</li> <li>Clean up all spills immediately.</li> <li>Avoid breathing vapours and contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> <li>Contain and absorb spill with sand, earth, inert material or vermiculite.</li> <li>Wipe up.</li> <li>Place in a suitable, labelled container for waste disposal.</li> </ul>
Major Spills	Moderate hazard.  Clear area of personnel and move upwind.  Alert Fire Brigade and tell them location and nature of hazard.  Wear breathing apparatus plus protective gloves.  Prevent, by any means available, spillage from entering drains or water course.  No smoking, naked lights or ignition sources.  Increase ventilation.  Stop leak if safe to do so.  Contain spill with sand, earth or vermiculite.  Collect recoverable product into labelled containers for recycling.  Absorb remaining product with sand, earth or vermiculite.  Collect solid residues and seal in labelled drums for disposal.  Wash area and prevent runoff into drains.  If contamination of drains or waterways occurs, advise emergency services.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

#### **SECTION 7 HANDLING AND STORAGE**

#### Precautions for safe handling

Precautions for safe handling		
Safe handling	<ul> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Prevent concentration in hollows and sumps.</li> <li>DO NOT enter confined spaces until atmosphere has been checked.</li> <li>Avoid smoking, naked lights or ignition sources.</li> <li>Avoid contact with incompatible materials.</li> <li>When handling, DO NOT eat, drink or smoke.</li> <li>Keep containers securely sealed when not in use.</li> <li>Avoid physical damage to containers.</li> <li>Always wash hands with soap and water after handling.</li> <li>Work clothes should be laundered separately.</li> <li>Use good occupational work practice.</li> </ul>	

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#### **Pink Detergent**

- Observe manufacturer's storage and handling recommendations contained within this SDS. Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions.
- ▶ DO NOT allow clothing wet with material to stay in contact with skin
- Store in original containers.
- Keep containers securely sealed.
- No smoking, naked lights or ignition sources.
- Other information Store in a cool, dry, well-ventilated area.
  - Store away from incompatible materials and foodstuff containers.
  - Protect containers against physical damage and check regularly for leaks.
  - ▶ Observe manufacturer's storage and handling recommendations contained within this SDS.

## Conditions for safe storage, including any incompatibilities

## Suitable container

- Packaging as recommended by manufacturer (HDPE).
- ► Check all containers are clearly labelled and free from leaks

Storage incompatibility

▶ Avoid reaction with oxidising agents

# SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

#### Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

#### INGREDIENT DATA

Not Available

#### **EMERGENCY LIMITS**

EMERICE CONTROL				
Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
Pink Detergent	Not Available	Not Available	Not Available	Not Available
Ingredient	Original IDLH		Revised IDLH	
(linear)alkylbenzenesulfonic acid, sodium salts	Not Available		Not Available	
diethanolamine cocoate	Not Available		Not Available	
sodium linear-(C12-14)alkyl ether sulfate	Not Available		Not Available	
1,2-benzisothiazoline-3-one	Not Available		Not Available	

#### MATERIAL DATA

for sodium hydroxide:

The TLV-C is recommended based on concentrations that produce noticeable but not excessive, ocular and upper respiratory tract irritation.

#### Exposure controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard 'physically' away from the worker and ventilation that strategically 'adds' and Appropriate engineering 'removes' air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match controls the particular process and chemical or contaminant in use

Employers may need to use multiple types of controls to prevent employee overexposure.

General exhaust is adequate under normal operating conditions. If risk of overexposure exists, wear SAA approved respirator. Correct fit is essential to obtain adequate protection. Provide adequate ventilation in warehouse or closed storage areas.

### Personal protection





# Eye and face protection

- Safety glasses with side shields
- Chemical goggles

Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

#### Skin protection

See Hand protection below

▶ Wear chemical protective gloves, e.g. PVC.

## NOTE:

The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid

#### Hands/feet protection

Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.

The selection of suitable gloves does not only depend on thematerial, but also on further marks of quality which vary from manufacturer tomanufacturer. Where the chemical is a preparation of several substances, theresistance of the glove material can not be calculated in advance and hastherefore to be checked prior to the application

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice

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Personal hygiene is a key element of effective hand care. Glovesmust only be worn on clean hands. After using gloves, hands should be washedand dried thoroughly. Application of a non-perfumed moisturizer is recommended.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:

- frequency and duration of contact,
- chemical resistance of glove material,
- glove thickness and
- dexterity
- Select gloves tested to a relevant standard (e.g. Europe EN 374,US F739, AS/NZS 2161.1 or national equivalent).
- When prolonged or frequently repeated contact may occur, a glovewith a protection class of 5 or higher (breakthrough time greater than 240minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) isrecommended.
- When only brief contact is expected, a glove with a protectionclass of 3 or higher (breakthrough time greater than 60 minutes according to EN374, AS/NZS 2161.10.1 or national equivalent) is recommended.
- Some glove polymer types are less affected by movement and thisshould be taken into account when considering gloves for long-term use.
- Contaminated gloves should be replaced.

For general applications, gloves with a thickness typicallygreater than 0.35 mm, are recommended.

It should be emphasised that glove thickness is not necessarily agood predictor of glove resistance to a specific chemical, as the permeationefficiency of the glove will be dependent on the exact composition of the glovematerial. Therefore, glove selection should also be based on consideration of the task requirements and knowledge of breakthrough times.

Glove thickness may also vary depending on the glove manufacturer, the glove type and the glove model. Therefore, the manufacturers' technical data should always be taken into account to ensure selection of the mostappropriate glove for the task.

Note: Depending on the activity being conducted, gloves of varyingthickness may be required for specific tasks. For example:

Thinner gloves (down to 0.1 mm or less) may be required where ahigh degree of manual dexterity is needed. However, these gloves are onlylikely to give short duration protection and would normally be just for singleuse applications, then disposed of.

Thicker gloves (up to 3 mm or more) may be required where there is a mechanical (as well as a chemical) risk i.e. where there is abrasion or puncture potential Gloves must only be worn on clean hands. After using gloves, handsshould be washed and dried thoroughly. Application of a non-perfumedmoisturiser is recommended.

Body protection	See Other protection below
Other protection	<ul><li>Barrier cream.</li><li>Skin cleansing cream.</li><li>Eye wash unit.</li></ul>

Thermal hazards Not Available

#### **SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES**

#### Information on basic physical and chemical properties

Appearance	Pink Opaque Liquid		
Physical state	Liquid	Relative density (Water = 1)	0.99 - 1.01
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	7-8	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Non Flammable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

#### **SECTION 10 STABILITY AND REACTIVITY**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

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#### **SECTION 11 TOXICOLOGICAL INFORMATION**

#### Information on toxicological effects

	•		
Inhaled	The material is not thought to produce adverse health effects or irritation of the Nevertheless, good hygiene practice requires that exposure be kept to a minim		
Ingestion	The material has <b>NOT</b> been classified by EC Directives or other classification animal or human evidence. The material may still be damaging to the health of kidney) damage is evident. Present definitions of harmful or toxic substances at morbidity (disease, ill-health). Gastrointestinal tract discomfort may produce na quantities is not thought to be cause for concern.	the individual, following ingere generally based on doses	estion, especially where pre-existing organ (e.g liver, producing mortality rather than those producing
Skin Contact	Evidence exists, or practical experience predicts, that the material either produces inflammation of the skin in a substantial number of individuals following direct contact, and/or produces significant inflammation when applied to the healthy intact skin of animals, for up to four hours, such inflammation being present twenty-four hours or more after the end of the exposure period. Skin irritation may also be present after prolonged or repeated exposure; this may result in a form of contact dermatitis (nonallergic). The dermatitis is often characterised by skin redness (erythema) and swelling (oedema) which may progress to blistering (vesiculation), scaling and thickening of the epidermis. At the microscopic level there may be intercellular oedema of the spongy layer of the skin (spongiosis) and intracellular oedema of the epidermis.  The material may accentuate any pre-existing dermatitis condition  Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions.		
Eye	When applied to the eye(s) of animals, the material produces severe ocular les	ions which are present twent	y-four hours or more after instillation.
Chronic	Practical experience shows that skin contact with the material is capable either of producing a positive response in experimental animals.  Repeated skin contact with some sulfonated surfactants has produced sensitis: There exists limited evidence that shows that skin contact with the material is c individuals, and/or of producing positive response in experimental animals.	ation dermatitis in predispose	ed individuals.
	TOXICITY IRRITATION		
Pink Detergent	Not Available Not Available		
(linear)alkylbenzenesulfonic	TOXICITY		IRRITATION
acid, sodium salts	Oral (rat) LD50: 800 mg/kg <sup>[2]</sup>		Nil reported
	TOXICITY	IRRITATION	
diethanolamine cocoate			
	Not Available	Not Available	
	TOXICITY		IRRITATION
sodium linear-(C12-14)alkyl ether sulfate	dermal (rat) LD50: >2000 mg/kg <sup>[1]</sup>		Not Available
etilei Sullate	Oral (rat) LD50: 4100 mg/kg <sup>[1]</sup>		
	TOXICITY	IRRITATION	
1,2-benzisothiazoline-3-one	Tan.		untation
1,2-56112150111102011116-5-0116	Oral (rat) LD50: 670 mg/kg <sup>[2]</sup> *MAK Documental		HILAUUTI
		Nil reported	
Legend:	Value obtained from Europe ECHA Registered Substances - Acute toxicity 2     Acute of chamical Substances	." value obtained from manu	ıracturer's SDS. Uniess otherwise specified data

Linear alkylbenzene sulfonates (LAS) are classified as Irritant (Xi) with the risk/hazard phrases R38/H315 (Irritating to skin) and R41/H318 (Risk of serious damage to eyes) according to CESIO (CESIO 2000). LAS are not included in Annex 1/6 of list of dangerous substances of Council Directive 67/548/EEC and

Linear alkylbenzene sulfonic acids (LABS) are strong acids (pKa<2) are classified as corrosive (R34/H314)

extracted from RTECS - Register of Toxic Effect of chemical Substances

1272/2008 respectively.

Acute toxicity: The available data indicate minimal to moderate toxicity, with LD50 values ranging from 500 to 2000 mg/kg body weight (bw). Acute inhalation data also indicate a lack of significant toxicity. Available dermal exposure data also shows a lack of significant toxicity.

LAS are readily absorbed by the gastrointestinal tract after oral administration in animals. LAS are not readily absorbed through the skin. The bulk is metabolised in the liver to sulfophenylic carboxyl acids. The metabolites are excreted primarily via the urine and faeces. The main urinary metabolites in rats are sulfophenyl butanoic acid and sulfophenyl pentanoic acid. Accumulation of LAS or its main metabolites has not been established in any organ after repeated oral ingestion.

#### Pink Detergent

No serious injuries or fatalities in man have been reported following accidental ingestion of LAS-containing detergent. The main clinical signs observed after oral administration to rats of doses near or greater than the LD50 values consisted of reduced voluntary activity, diarrhoea, weakness etc. Death usually occurred within 24 hours of administration. Rats appear to be more sensitive to LAS than mice.

LAS and branched alkylbenzene sulfonates may cause irritation of the eyes, skin and mucous membranes. LAS are relatively more irritating to the skin than the corresponding branched alkylbenzene sulfonates. The potential of LAS to irritate the skin depends on the concentration applied. LAS have been classified as irritating to skin at concentrations above 20% according to EU-criteria. Human skin can tolerate contact with solution of up to 1% LAS for 24 hours resulting in only mild irritation. Application of > 5% LAS to the eyes of rabbits produced irritation. Concentration of < 0.1% LAS produced mild to no irritation. Skin sensitization was not seen in 2.294 volunteers exposed to LAS or in 17.887 exposed to formulations of LAS.

Repeat dose toxicity: A feeding study indicated that LAS, when administered for 2 years at extremely high levels (0.5%) in the diets to rats, produced no adverse effects on growth, health or feed efficiency.

**Genotoxicity:** The mutagenic potential of LAS was tested using *Salmonella typhimurium* strains, using Ames test. In these studies, LAS was not mutagenic. The available long-term studies are inadequate for evaluating the carcinogenic potential of LAS in laboratory animals. The studies available (oral

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administration to rats and mice) do not show any evidence of carcinogenicity.

Reproductive toxicity: In general no specific effect of LAS on reproductive processes has been seen, although dosages causing maternal toxicity may also induce some effects on reproduction. No teratogenic effects attributed to LAS exposure have been observed.

Environmental and Health Assessment of Substances in Household Detergents and Cosmetic Detergent Products, Environment Project, 615, 2001. Torben Madsen et al: Milioministeriet (Danish Environmental Protection Agency)

Acute Toxicity	0	Carcinogenicity	0
Skin Irritation/Corrosion	✓	Reproductivity	0
Serious Eye Damage/Irritation	<b>~</b>	STOT - Single Exposure	0
Respiratory or Skin sensitisation	0	STOT - Repeated Exposure	0
Mutagenicity	0	Aspiration Hazard	0

Leaend:

💢 – Data available but does not fill the criteria for classification

Data required to make classification available

Data Not Available to make classification

#### **SECTION 12 ECOLOGICAL INFORMATION**

#### **Toxicity**

Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
(linear)alkylbenzenesulfonic acid, sodium salts	LC50	96	Fish	5mg/L	4
(linear)alkylbenzenesulfonic acid, sodium salts	EC50	120	Algae or other aquatic plants	10mg/L	4
diethanolamine cocoate	LC50	96	Fish	=2.8mg/L	1
diethanolamine cocoate	EC50	48	Crustacea	=2.39mg/L	1
diethanolamine cocoate	EC50	96	Algae or other aquatic plants	=2.3mg/L	1
diethanolamine cocoate	EC0	96	Algae or other aquatic plants	=1mg/L	1
diethanolamine cocoate	NOEC	504	Crustacea	=1mg/L	1
sodium linear-(C12-14)alkyl ether sulfate	LC50	96	Fish	>1- <10mg/L	2
sodium linear-(C12-14)alkyl ether sulfate	EC50	48	Crustacea	7.2mg/L	2
sodium linear-(C12-14)alkyl ether sulfate	EC50	72	Algae or other aquatic plants	1.8mg/L	2
sodium linear-(C12-14)alkyl ether sulfate	EC50	504	Crustacea	0.37mg/L	2
sodium linear-(C12-14)alkyl ether sulfate	NOEC	672	Fish	0.1mg/L	2
1,2-benzisothiazoline-3-one	LC50	96	Fish	1.6mg/L	4
1,2-benzisothiazoline-3-one	EC50	48	Crustacea	0.062mg/L	4
1,2-benzisothiazoline-3-one	EC50	48	Crustacea	4.4mg/L	4
Legend:	Aquatic Toxicity Data (E.	,	nistered Substances - Ecotoxicological Informat se - Aquatic Toxicity Data 5. ECETOC Aquatic I ata 8. Vendor Data	, ,	

#### For surfactants:

#### **Environmental fate:**

Octanol/water partition coefficients cannot easily be determined for surfactants because one part of the molecule is hydrophilic and the other part is hydrophobic. Consequently they tend to accumulate at the interface and are not extracted into one or other of the liquid phases. As a result surfactants are expected to transfer slowly, for example, from water into the flesh of fish. During this process, readily biodegradable surfactants are expected to be metabolised rapidly during the process of bioaccumulation. This was emphasised by the OECD Expert Group stating that chemicals are not to be considered to show bioaccumulation potential if they are readily biodegradable.

Several anionic and nonionic surfactants have been investigated to evaluate their potential to bioconcentrate in fish. BCF values (BCF - bioconcentration factor) ranging from 1 to 350 were found. These are absolute maximum values, resulting from the radiolabelling technique used. In all these studies, substantial oxidative metabolism was found resulting in the highest radioactivity in the gall bladder. This indicates liver transformation of the parent compound and biliary excretion of the metabolised compounds, so that 'real' bioconcentration is overstated. After correction it can be expected that 'real' parent BCF values are one order of magnitude less than those indicated above, i.e. 'real' BCF is <100. Therefore the usual data used for classification by EU directives to determine whether a substance is 'Dangerous to the 'Environment' has little bearing on whether the use of the surfactant is environmentally acceptable

Surfactant should be considered to be toxic (EC50 and LC50 values of < 10 mg/L) to aquatic species under conditions that allow contact of the chemicals with the organisms. The water solubility of the chemicals does not impact the toxicity except as it relates to the ability to conduct tests appropriately to obtain exposure of the test species. The acute aquatic toxicity generally is considered to be related to the effects of the surfactant properties on the organism and not to direct chemical toxicity

#### For linear alkylbenzene sulfonic acids (LABS) (and their salts): **Environmental fate:**

LABS are highly water soluble (miscible) and have a relatively lowKow. The environmental fate data indicate that these chemicals are highly susceptible to photo-and biodegradation. LABS are strong acids (pKa <1) that are completelyionised in aqueous solutions. The chemical species present in aqueous solutionsat neutral (physiological) pH is the linear alkylbenzene sulfonate (the LASion) (C10-14 linear alkyl benzene-SO3-), the identical species present insolutions of LAS, where the counter ion (typically sodium, calcium or ammonium) will disassociate to form the LAS anion. Thus, the physical-chemical environmental fate, ecotoxicity and toxicity properties of the LABS and LASwould be expected to be similar. It should be noted that the LABS are liquidsand LAS is a solid at room temperature. However, in water the difference betweenthe LAB sulfonic acids and LAS disappears as dissociation results in the sameion in solution. Therefore, parameters such as Kow, water solubility and pH/pKaare appropriate to compare. The octanol-water partition coefficients are around2 (logKow) for all of the chemicals in this category LABS are not expected to volatilise significantly. Fugacitymodeling predicts that most of these chemicals will partition to the soil andwater. Very little partitions to the air or sediment. Photodegradation is estimated (using EPI Suite software) to be asignificant mechanism for breakdown. Based on the model estimates, the hydroxylradical reaction half-lives ranged from about 7 to 8.6 hours. Estimated datafor LAS were similar. Furthermore, measured data for LAS suggest even morerapid photodegradation, with 95% of the material degraded within 20 minutes at 20 C in a laboratory study.

Experimental data data indicates that LAS is stable in water.

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## **Pink Detergent**

LABS are generally biodegradable. Measured biodegradation data indicatesubstantial microbial degradation under aerobic conditions. For dodecylbenzenesulfonic acid 69% of the material mineralised after 28 days. Biodegradation ofthe C10-16 derivatives and the LAS are also rapid, with 93% or greater of thematerial degrading within 28 or 37 days. In addition, studies show thatstraight chain alkylbenzene sulfonate materials readily degrade, with theshorter chain length compounds degrading more rapidly. Thus, the dataindicate that these chemicals are highly susceptible to degradation, both byphotolytic and microbial mechanisms

The initial step in the biodegradation of LABS under aerobicconditions is an omega -oxidation of the terminal methyl group of the alkylchain to form a carboxylic acid. Further degradation proceeds by a stepwiseshortening of the alkyl chain by beta -oxidation leaving a short-chainsulfophenyl carboxylic acid. In the presence of molecular oxygen the aromaticing structure hydrolyses to form a dihydroxy-benzene structure which is openedbefore desulfonation of the formed sulfonated dicarboxylic acid. The finaldegradation steps have not been investigated in details but are likely to occurby general bacterial metabolic routes involving a total mineralisation andassimilation into biomass. Both the initial omega -oxidation and thehydroxylation of the ring structure of LAS require molecular oxygen, and theyare not expected to take place under anoxic conditions.

The BioConcentration Factor (BCF) tends to increase withincreasing alkyl chain length but also the position of the aryl sulfonatemoiety was important. A higher BCF was seen for linear alkyl benzenesulfonateisomers with the aryl sulfonate attached. Available data indicate that LABShave low to moderate bioaccumulation potential, with a bioconcentration factor of dodecyl benzene sulfonic acid of 130. LAS has bioconcentration factors thatrange from 22 to 87.

#### Ecotoxicity:

Numerous studies have been performed to determine the effects of LABS towards aquatic organisms. The aquatic effect concentrations that wereobserved in these studies are highly variable. This variation is partly related to the testing of different isomers and homologues, but it may also be due to the specific test conditions and species. The length of the alkyl chain is animportant factor determining the aquatic toxicity. In general, the homologues with the highest number of carbons in the alkyl chain are more toxic than arethose with shorter alkyl chains. Today, commercial LABS have a homologuedistribution between C10 and C13 with a typical average alkyl chain length of C11.6.

The widest range in the toxicity of LABS towards species belonging to the same group is found for algae Approximately 90% of the data found in the literature fall between 0.1 and 100 mg/l. Typical ranges of EC50 valuesare 1 to 100 mg/l for fresh water species and < 1 to 10 mg/l for marinespecies. Typical values lie between 29 and 170 mg/l

A very low EC100 value of 0.025 mg/l was determined for *Gymnodiumbreve*. Previous studies in which *Gymnodium breve* wasexposed with AES confirm that this species is highly sensitive to surfactants, and occasionally available data for *Gymnodium breve* shouldtherefore not be used for comparison of the aquatic toxicity between varioussurfactants.

LC50 values have been found in the range of 1 to 10 mg/lwhen Daphnia magna were exposed with LABS homologues between C10 and C13. The acute toxicity of LABS to Daphnia magna generally increases with increasing alkyl chain length. Typical values lie between 3 and 12 mg/l.

A study with the marine crustacean Acartia tonsa indicated that a C10-13 LAS affected the survival at 0.54 mg/l (LC50) and the development at 0.51 mg/l (EC50) after 8 days of exposure. The 48 h-LC50 that was obtained in the same study with Acartia tonsa was 2.1 mg/l.

Metabolites from biotransformation of LABS are reported to have amuch lower toxicity to invertebrates compared to the toxicity of the intactsurfactant.

The toxicity of LABS to fish generally increases with increasingalkyl chain length, and approximately a 10-fold difference in toxicity betweenhomologues separated by two carbon atoms has been observed. As also noted forinvertebrates, fish are less susceptible to metabolites from biotransformation LABS. LC50 values below 1 mg/l were found for C11.9 (0.71 mg/l), C13 and C14 (both 0.4 mg/l) in studies with fathead minnow.

LABS sorb to sediment with partition coefficients of 50 to 1,000. The toxicity of LABS bound to sediment is relatively low compared to LABS insolution. NOEC and LOEC values were as high as 319 and 993 mg LABS/kg, respectively, for the sediment-living *Chironomus riparius*. The corresponding NOEC for LABS in solution was as low as 2.4 mg/l indicating thatonly a small fraction of the sorbed LABS was bioavailable. LABS dissolved inwater may also cause chronic effects like reduction of the growth rate of themarine mussel *Mytilus galloprovincialis*. LABS sorbed to sedimentsdid not have similar effects.

Environmental and Health Assessment of Substances in HouseholdDetergents and Cosmetic Detergent Products, Environment Project, 615, 2001. Torben Madsen et al: Miljoministeriet (Danish Environmental ProtectionAgency) Assessment Plan for the Linear Alkylbenzene (LAB) Sulfonic AcidsCategory in Accordance with the USEPA High Production Volume Chemical Challenge Program: The LAB Sulfonic Acids Coalition

#### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
	No Data available for all ingredients	No Data available for all ingredients

#### Bioaccumulative potential

Ingredient	Bioaccumulation
	No Data available for all ingredients

#### Mobility in soil

Ingredient	Mobility
	No Data available for all ingredients

#### **SECTION 13 DISPOSAL CONSIDERATIONS**

#### Waste treatment methods

Product / Packaging

disposa

- ► Containers may still present a chemical hazard/ danger when empty.
- ► Return to supplier for reuse/ recycling if possible.

#### Otherwise:

- If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
- ▶ Where possible retain label warnings and SDS and observe all notices pertaining to the product.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- ▶ Reduction
  - ▶ Reuse
  - ▶ Recycling
  - ▶ Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

- ► DO NOT allow wash water from cleaning or process equipment to enter drains
- It may be necessary to collect all wash water for treatment before disposal.
- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority.
- Recycle wherever possible or consult manufacturer for recycling options.
- Consult State Land Waste Authority for disposal.
- ▶ Bury or incinerate residue at an approved site.
- ▶ Recycle containers if possible, or dispose of in an authorised landfill.

#### **SECTION 14 TRANSPORT INFORMATION**

## **Pink Detergent**

**Labels Required** 

**Marine Pollutant** NO **HAZCHEM** Not Applicable

Land transport (ADG): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

#### **SECTION 15 REGULATORY INFORMATION**

Safety, health and environmental regulations / legislation specific for the substance or mixture

(LINEAR)ALKYLBENZENESULFONIC ACID, SODIUM SALTS(8046-53-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Not Applicable

DIETHANOLAMINE COCOATE(8051-30-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Inventory of Chemical Substances (AICS)

SODIUM LINEAR-(C12-14)ALKYL ETHER SULFATE(68891-38-3) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Hazardous Substances Information System - Consolidated Lists

Australia Inventory of Chemical Substances (AICS)

#### 1,2-BENZISOTHIAZOLINE-3-ONE(2634-33-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Hazardous Substances	ces Information System - Consolidated Lists Australia Inventory of Chemical Su	ubstances (AICS)
National Inventory	Status	
Australia - AICS	N ((linear)alkylbenzenesulfonic acid, sodium salts)	
Canada - DSL	N ((linear)alkylbenzenesulfonic acid, sodium salts)	
Canada - NDSL	N (diethanolamine cocoate; sodium linear-(C12-14)alkyl ether sulfate; 1,2-benzisothiazoline-3-one; (linear)alkylbenzenesulfonic acid, sodium salts)	
China - IECSC	Υ	
Europe - EINEC / ELINCS / NLP	N ((linear)alkylbenzenesulfonic acid, sodium salts)	
Japan - ENCS	N ((linear)alkylbenzenesulfonic acid, sodium salts)	
Korea - KECI	N ((linear)alkylbenzenesulfonic acid, sodium salts)	
New Zealand - NZIoC	N ((linear)alkylbenzenesulfonic acid, sodium salts)	
Philippines - PICCS	N ((linear)alkylbenzenesulfonic acid, sodium salts)	
USA - TSCA	N ((linear)alkylbenzenesulfonic acid, sodium salts)	
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)	

#### **SECTION 16 OTHER INFORMATION**

#### Other information

#### Ingredients with multiple cas numbers

Name	CAS No
sodium linear-(C12-14)alkyl ether sulfate	68891-38-3, 9004-82-4

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### **Definitions and abbreviations**

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit,

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection

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# **Pink Detergent**

OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

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